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Review Module / Chapters 13–16 7 In your notebooks, solve the following problems.

SECTION 16.1 THE NATURE OF COVALENT BONDING 1. Draw the electron dot structure for hydrogen fluoride, HF. 2. Draw the electron dot structure for phosphorus trifluoride, PF₃. 3. Draw the electron dot structure for nitrogen trichloride. 4.

16 Covalent Bonding Practice Problems

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