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own symbol, which is a capital letter M. A solution that has the concentration 5 mol/L would be called a 5 M solution or said to have a concentration value of 5 molar.

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where the subscripts n_1 and n_2 refer to
the solution before and after the dilution,
respectively. Since the dilution process
does not change the amount of solute in
the solution, $n_1 = n_2$. Thus, these two
equations may be set equal to one another:
 $[M_1L_1 = M_2L_2]$ This relation is
commonly referred to as the dilution
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