

Chapter 17 Acids Bases Worksheet

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17.1 Buffers

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Grade 2 Module 3 Lesson 17 Acid-Base Equilibria and Buffer Solutions

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17 Acids and Bases (same as Exam 2 worksheet)

for electron-withdrawing groups, as electronegativity increases, acidity increases. -as the number of oxygen atoms attached to Y (central atom) increases, acidity increases. Strong Acids. -acids that completely ionize in aqueous solution. 1. HCL (Hydrochloric acid) 2. HI (Hydroiodic acid) 3.

Chapter 17: Acids and Bases Flashcards | Quizlet

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1. Mix together a weak acid and its conjugate base or vice versa 2. Mix an excess amount of weak acid with a limiting amount of strong base (to produce the conjugate weak base, with excess weak acid) 3. Mix an excess amount of weak bases with a limiting amount of strong acid (to produce the conjugate weak acid, with excess weak base left over)